

2-Benzyl-4-(phenylsulfonyl)-1,2,3,4-tetrahydropyrrolo[3,4-*b*]indole

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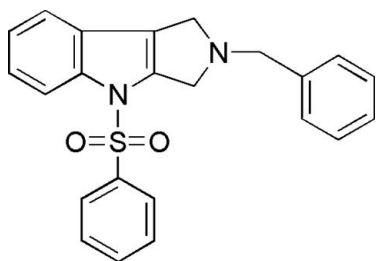
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.040; wR factor = 0.117; data-to-parameter ratio = 17.0.

The indole and pyrrolidine ring systems of the title compound, $\text{C}_{23}\text{H}_{20}\text{N}_2\text{O}_2\text{S}$, are essentially coplanar. The angle between the planes of the phenylsulfonyl group and the indole ring system is 77.0 (1)°. The benzyl ring and the pyrroloindole plane are nearly perpendicular, with an angle between the planes of 102.0 (1)°.

Related literature

For the synthesis of the title compound and related chemistry, see: Gribble (2003); Gribble *et al.* (2005); Roy, Kishbaugh *et al.* (2007); Roy, Pelkey *et al.* (2007); Kishbaugh & Gribble (2002); Mohanakrishnan & Srinivasan (1995). For a related structure, see: Kishbaugh *et al.* (2007).



Experimental

Crystal data

$\text{C}_{23}\text{H}_{20}\text{N}_2\text{O}_2\text{S}$
 $M_r = 388.47$
Triclinic, $P\bar{1}$

$a = 9.178$ (2) Å
 $b = 9.6463$ (17) Å
 $c = 12.0124$ (14) Å

$\alpha = 78.680$ (12)°
 $\beta = 110.275$ (13)°
 $\gamma = 107.904$ (18)°
 $V = 944.5$ (3) Å³
 $Z = 2$

Mo $K\alpha$ radiation
 $\mu = 0.19$ mm⁻¹
 $T = 296$ K
 $0.50 \times 0.30 \times 0.25$ mm

Data collection

Rigaku AFC-6S diffractometer
Absorption correction: ψ scan
(North *et al.*, 1968)
 $T_{\min} = 0.910$, $T_{\max} = 0.953$
4591 measured reflections
4328 independent reflections

2280 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$
3 standard reflections
every 150 reflections
intensity decay: none

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.117$
 $S = 0.99$
4328 reflections

255 parameters
H-atom parameters constrained
 $\Delta\rho_{\max} = 0.28$ e Å⁻³
 $\Delta\rho_{\min} = -0.31$ e Å⁻³

Data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1994); cell refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *CrystalStructure* (Rigaku/MSC, 2005); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *CrystalStructure*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FL2138).

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supplementary materials

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2-Benzyl-4-(phenylsulfonyl)-1,2,3,4-tetrahydropyrrolo[3,4-*b*]indole

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Comment

The title compound, C₂₃H₂₀N₂O₂S, was synthesized as a precursor to the corresponding pyrrolo[3,4-*b*]indole as a stable synthetic analogue of indole-2,3-quinodimethane and having a potentially removable benzyl protecting group. The X-ray crystallographic analysis confirms the molecular structure and atom connectivity for (I) that we had proposed for this compound based on NMR spectroscopy and chemical reactions (Kishbaugh & Gribble, 2002).

The pyrrolidine nitrogen N2 has bond angles totaling 336.4 (2)°, indicating slight flattening from a purely pyramidal nitrogen, which is 324° for ammonia with H—N—H bond angles of 108°, and is consistent with the inductive electron-withdrawing properties of a benzyl group.

Experimental

To a refluxing solution of 2,3-dibromomethyl-1-(phenylsulfonyl)indole (668 mg, 1.51 mmol) (Mohanakrishnan & Srinivasan, 1995) and K₂CO₃ (625 mg, 4.50 mmol) in tetrahydrofuran (20 ml) was added a solution of benzylamine (200 μL, 1.80 mmol) in tetrahydrofuran (20 ml) slowly *via* addition funnel. After 14 h, the opaque solution was filtered through a Celite pad with ethyl acetate rinses. The combined yellow filtrate was concentrated *in vacuo* to yield a (I) as a pale yellow solid, which was purified by column chromatography (3:1 hexanes: ethyl acetate) to yield (I) (360 mg) in 61% yield: m.p. 424–425 K. As (I) was unstable toward oxidation, an elemental analysis was not attempted. Recrystallization from hexane–dichloromethane (3:1) yielded crystals that were suitable for X-ray crystallography.

Refinement

The H atoms were included in the riding model approximation with C—H = 0.93–0.97 Å, and with $U_{\text{iso}}(\text{H}) = 1.19\text{--}1.20U_{\text{eq}}(\text{C})$.

Figures

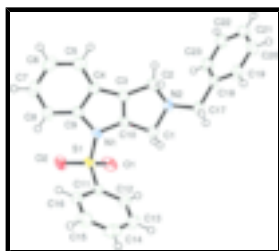


Fig. 1. Molecular structure of (I) showing atom labelling and 50% probability displacement ellipsoids.

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Crystal data

$C_{23}H_{20}N_2O_2S$	$Z = 2$
$M_r = 388.47$	$F_{000} = 408$
Triclinic, $P\bar{1}$	$D_x = 1.366 \text{ Mg m}^{-3}$
$a = 9.178 (2) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 9.6463 (17) \text{ \AA}$	$\lambda = 0.71069 \text{ \AA}$
$c = 12.0124 (14) \text{ \AA}$	Cell parameters from 25 reflections
$\alpha = 78.680 (12)^\circ$	$\theta = 11.2\text{--}19.2^\circ$
$\beta = 110.275 (13)^\circ$	$\mu = 0.19 \text{ mm}^{-1}$
$\gamma = 107.904 (18)^\circ$	$T = 296 \text{ K}$
$V = 944.5 (3) \text{ \AA}^3$	Prism, colourless
	$0.50 \times 0.30 \times 0.25 \text{ mm}$

Data collection

Rigaku AFC-6S diffractometer	$R_{\text{int}} = 0.042$
Radiation source: normal-focus sealed tube	$\theta_{\text{max}} = 27.5^\circ$
Monochromator: graphite	$\theta_{\text{min}} = 1.8^\circ$
$T = 296 \text{ K}$	$h = -11 \rightarrow 11$
$\omega/2\theta$ scans	$k = 0 \rightarrow 12$
Absorption correction: ψ scan (North <i>et al.</i> , 1968)	$l = -15 \rightarrow 15$
$T_{\text{min}} = 0.910$, $T_{\text{max}} = 0.953$	3 standard reflections
4591 measured reflections	every 150 reflections
4328 independent reflections	intensity decay: none
2280 reflections with $I > 2\sigma(I)$	

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.040$	$w = 1/[\sigma^2(F_o^2) + (0.0326P)^2 + 0.5372P]$
$wR(F^2) = 0.117$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 0.99$	$(\Delta/\sigma)_{\text{max}} < 0.001$
4328 reflections	$\Delta\rho_{\text{max}} = 0.28 \text{ e \AA}^{-3}$
255 parameters	$\Delta\rho_{\text{min}} = -0.31 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: SHELXL97 (Sheldrick, 1997), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
Secondary atom site location: difference Fourier map	Extinction coefficient: 0.0159 (16)

Special details

Experimental. ^1H (CD_2Cl_2) δ 8.01 (dd, 1H, 1, 8 Hz), 7.85–7.88 (m, 2H), 7.22–7.57 (m, 11H), 4.29 (m, 2H) 4.05 (s, 2H), 3.93 (m, 2H); ^{13}C (CD_2Cl_2) (500 MHz) δ 140.2, 139.8, 139.5, 138.4, 134.3, 129.7, 129.3, 129.0, 128.8, 128.7, 128.6, 127.5, 127.0, 126.7, 126.3, 124.03, 124.02, 123.7, 119.6, 114.5, 60.8, 55.1, 53.2; IR (film) λ_{max} 3055, 2800, 1448, 1369, 1175, 1094, 997, 749, 685 cm^{-1} ; UV (EtOH) λ_{max} 260 nm.

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\text{sigma}(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.22121 (8)	0.47346 (7)	0.12134 (6)	0.03143 (18)
O1	0.0586 (2)	0.4125 (2)	0.12119 (16)	0.0438 (5)
O2	0.2746 (2)	0.4548 (2)	0.02823 (15)	0.0452 (5)
N1	0.3320 (3)	0.3986 (2)	0.24764 (17)	0.0304 (5)
N2	0.1871 (3)	0.2885 (2)	0.50729 (18)	0.0339 (5)
C1	0.1751 (3)	0.3952 (3)	0.4002 (2)	0.0349 (6)
H1	0.2078	0.4952	0.4203	0.042*
H2	0.0674	0.3756	0.3439	0.042*
C2	0.3484 (3)	0.2609 (3)	0.5499 (2)	0.0386 (7)
H3	0.3439	0.1601	0.5832	0.046*
H4	0.4234	0.3258	0.6089	0.046*
C3	0.3927 (3)	0.2933 (3)	0.4369 (2)	0.0322 (6)
C4	0.5080 (3)	0.2795 (3)	0.3867 (2)	0.0314 (6)
C5	0.6387 (3)	0.2191 (3)	0.4284 (3)	0.0421 (7)
H5	0.6660	0.1736	0.5057	0.050*
C6	0.7274 (3)	0.2269 (3)	0.3547 (3)	0.0457 (7)
H6	0.8141	0.1853	0.3823	0.055*
C7	0.6890 (3)	0.2964 (3)	0.2391 (3)	0.0443 (7)
H7	0.7526	0.3025	0.1917	0.053*
C8	0.5592 (3)	0.3563 (3)	0.1935 (2)	0.0360 (6)
H8	0.5330	0.4010	0.1158	0.043*
C9	0.4691 (3)	0.3474 (3)	0.2679 (2)	0.0291 (6)
C10	0.2923 (3)	0.3644 (3)	0.3543 (2)	0.0288 (6)
C11	0.2788 (3)	0.6593 (3)	0.1432 (2)	0.0285 (6)
C12	0.1915 (3)	0.7117 (3)	0.1910 (2)	0.0396 (7)
H9	0.0985	0.6509	0.2063	0.048*
C13	0.2448 (4)	0.8561 (3)	0.2159 (3)	0.0463 (8)

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H10	0.1876	0.8929	0.2484	0.056*
C14	0.3820 (4)	0.9452 (3)	0.1928 (3)	0.0492 (8)
H11	0.4173	1.0420	0.2103	0.059*
C15	0.4678 (4)	0.8933 (3)	0.1442 (3)	0.0465 (7)
H12	0.5601	0.9550	0.1285	0.056*
C16	0.4166 (3)	0.7493 (3)	0.1188 (2)	0.0370 (6)
H13	0.4738	0.7133	0.0857	0.044*
C17	0.1417 (3)	0.3296 (3)	0.5993 (2)	0.0369 (6)
H19	0.0399	0.3553	0.5631	0.044*
H20	0.2222	0.4153	0.6329	0.044*
C18	0.1263 (3)	0.2072 (3)	0.6980 (2)	0.0307 (6)
C19	0.1778 (3)	0.2378 (3)	0.8158 (2)	0.0349 (6)
H14	0.2230	0.3343	0.8338	0.042*
C20	0.1631 (3)	0.1271 (3)	0.9067 (2)	0.0427 (7)
H15	0.1973	0.1498	0.9849	0.051*
C21	0.0981 (4)	-0.0168 (3)	0.8823 (3)	0.0460 (7)
H16	0.0902	-0.0915	0.9437	0.055*
C22	0.0445 (3)	-0.0486 (3)	0.7648 (3)	0.0422 (7)
H17	-0.0006	-0.1452	0.7471	0.051*
C23	0.0578 (3)	0.0622 (3)	0.6743 (2)	0.0408 (7)
H18	0.0202	0.0394	0.5959	0.049*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0382 (4)	0.0303 (4)	0.0217 (3)	0.0064 (3)	0.0077 (3)	-0.0002 (2)
O1	0.0308 (11)	0.0450 (12)	0.0380 (11)	-0.0015 (9)	0.0027 (8)	0.0027 (9)
O2	0.0691 (14)	0.0461 (12)	0.0249 (10)	0.0171 (11)	0.0175 (10)	-0.0046 (9)
N1	0.0391 (13)	0.0323 (12)	0.0222 (11)	0.0127 (10)	0.0119 (9)	0.0020 (9)
N2	0.0394 (13)	0.0411 (13)	0.0258 (11)	0.0136 (11)	0.0158 (10)	0.0016 (9)
C1	0.0383 (15)	0.0403 (16)	0.0278 (14)	0.0129 (13)	0.0121 (12)	0.0007 (11)
C2	0.0420 (16)	0.0491 (18)	0.0250 (13)	0.0159 (14)	0.0120 (12)	0.0050 (12)
C3	0.0316 (14)	0.0388 (15)	0.0251 (13)	0.0088 (12)	0.0099 (11)	0.0011 (11)
C4	0.0303 (14)	0.0333 (14)	0.0299 (14)	0.0078 (11)	0.0100 (11)	-0.0009 (11)
C5	0.0379 (16)	0.0510 (18)	0.0379 (16)	0.0177 (14)	0.0099 (13)	0.0012 (13)
C6	0.0345 (16)	0.0547 (19)	0.0516 (19)	0.0172 (14)	0.0120 (14)	-0.0067 (15)
C7	0.0403 (17)	0.0516 (19)	0.0522 (19)	0.0088 (14)	0.0268 (15)	-0.0124 (15)
C8	0.0395 (16)	0.0376 (15)	0.0344 (15)	0.0062 (13)	0.0186 (12)	-0.0041 (12)
C9	0.0315 (14)	0.0255 (13)	0.0295 (13)	0.0050 (11)	0.0104 (11)	-0.0031 (10)
C10	0.0316 (14)	0.0334 (14)	0.0220 (12)	0.0078 (11)	0.0108 (11)	-0.0004 (10)
C11	0.0305 (14)	0.0291 (13)	0.0254 (13)	0.0098 (11)	0.0089 (11)	0.0021 (10)
C12	0.0395 (16)	0.0400 (17)	0.0441 (16)	0.0129 (13)	0.0200 (13)	0.0024 (13)
C13	0.0523 (19)	0.0469 (19)	0.0499 (18)	0.0258 (16)	0.0171 (15)	-0.0042 (14)
C14	0.054 (2)	0.0346 (17)	0.0497 (19)	0.0118 (15)	0.0022 (16)	-0.0084 (14)
C15	0.0393 (17)	0.0381 (17)	0.0543 (19)	0.0001 (13)	0.0156 (15)	-0.0009 (14)
C16	0.0359 (16)	0.0387 (16)	0.0386 (15)	0.0082 (13)	0.0169 (13)	-0.0016 (12)
C17	0.0444 (17)	0.0386 (16)	0.0337 (15)	0.0101 (13)	0.0198 (13)	-0.0037 (12)
C18	0.0334 (14)	0.0355 (14)	0.0290 (13)	0.0076 (12)	0.0171 (11)	-0.0047 (11)

C19	0.0355 (15)	0.0353 (15)	0.0353 (15)	0.0031 (12)	0.0152 (12)	-0.0086 (12)
C20	0.0469 (18)	0.0547 (19)	0.0278 (14)	0.0116 (15)	0.0145 (13)	-0.0039 (13)
C21	0.0520 (19)	0.0474 (19)	0.0445 (18)	0.0180 (15)	0.0261 (15)	0.0108 (14)
C22	0.0503 (18)	0.0301 (15)	0.0536 (18)	0.0051 (13)	0.0277 (15)	-0.0073 (13)
C23	0.0529 (18)	0.0373 (16)	0.0352 (15)	0.0037 (14)	0.0200 (14)	-0.0111 (12)

Geometric parameters (Å, °)

S1—O1	1.4259 (19)	C8—H8	0.9300
S1—O2	1.4271 (19)	C11—C12	1.382 (4)
S1—N1	1.676 (2)	C11—C16	1.386 (3)
S1—C11	1.751 (3)	C12—C13	1.383 (4)
N1—C10	1.402 (3)	C12—H9	0.9300
N1—C9	1.417 (3)	C13—C14	1.372 (4)
N2—C17	1.464 (3)	C13—H10	0.9300
N2—C1	1.476 (3)	C14—C15	1.373 (4)
N2—C2	1.480 (3)	C14—H11	0.9300
C1—C10	1.491 (3)	C15—C16	1.382 (4)
C1—H1	0.9700	C15—H12	0.9300
C1—H2	0.9700	C16—H13	0.9300
C2—C3	1.497 (3)	C17—C18	1.512 (3)
C2—H3	0.9700	C17—H19	0.9700
C2—H4	0.9700	C17—H20	0.9700
C3—C10	1.338 (3)	C18—C23	1.388 (4)
C3—C4	1.436 (3)	C18—C19	1.388 (3)
C4—C5	1.388 (4)	C19—C20	1.380 (4)
C4—C9	1.420 (3)	C19—H14	0.9300
C5—C6	1.373 (4)	C20—C21	1.377 (4)
C5—H5	0.9300	C20—H15	0.9300
C6—C7	1.393 (4)	C21—C22	1.387 (4)
C6—H6	0.9300	C21—H16	0.9300
C7—C8	1.379 (4)	C22—C23	1.376 (4)
C7—H7	0.9300	C22—H17	0.9300
C8—C9	1.389 (3)	C23—H18	0.9300
O1—S1—O2	121.61 (13)	C3—C10—N1	110.5 (2)
O1—S1—N1	105.35 (11)	C3—C10—C1	113.1 (2)
O2—S1—N1	106.06 (11)	N1—C10—C1	136.3 (2)
O1—S1—C11	109.19 (12)	C12—C11—C16	121.1 (2)
O2—S1—C11	109.65 (12)	C12—C11—S1	119.4 (2)
N1—S1—C11	103.31 (11)	C16—C11—S1	119.4 (2)
C10—N1—C9	106.93 (19)	C11—C12—C13	118.9 (3)
C10—N1—S1	123.40 (17)	C11—C12—H9	120.6
C9—N1—S1	129.48 (17)	C13—C12—H9	120.6
C17—N2—C1	113.0 (2)	C14—C13—C12	120.2 (3)
C17—N2—C2	114.1 (2)	C14—C13—H10	119.9
C1—N2—C2	109.32 (19)	C12—C13—H10	119.9
N2—C1—C10	99.3 (2)	C13—C14—C15	120.9 (3)
N2—C1—H1	111.9	C13—C14—H11	119.6
C10—C1—H1	111.9	C15—C14—H11	119.6

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N2—C1—H2	111.9	C14—C15—C16	119.9 (3)
C10—C1—H2	111.9	C14—C15—H12	120.1
H1—C1—H2	109.6	C16—C15—H12	120.1
N2—C2—C3	101.5 (2)	C15—C16—C11	119.1 (3)
N2—C2—H3	111.5	C15—C16—H13	120.5
C3—C2—H3	111.5	C11—C16—H13	120.5
N2—C2—H4	111.5	N2—C17—C18	112.0 (2)
C3—C2—H4	111.5	N2—C17—H19	109.2
H3—C2—H4	109.3	C18—C17—H19	109.2
C10—C3—C4	108.6 (2)	N2—C17—H20	109.2
C10—C3—C2	109.0 (2)	C18—C17—H20	109.2
C4—C3—C2	142.4 (2)	H19—C17—H20	107.9
C5—C4—C9	118.7 (2)	C23—C18—C19	117.9 (2)
C5—C4—C3	134.9 (2)	C23—C18—C17	121.6 (2)
C9—C4—C3	106.4 (2)	C19—C18—C17	120.5 (2)
C6—C5—C4	119.7 (3)	C20—C19—C18	121.0 (3)
C6—C5—H5	120.2	C20—C19—H14	119.5
C4—C5—H5	120.2	C18—C19—H14	119.5
C5—C6—C7	120.8 (3)	C21—C20—C19	120.5 (3)
C5—C6—H6	119.6	C21—C20—H15	119.7
C7—C6—H6	119.6	C19—C20—H15	119.7
C8—C7—C6	121.6 (3)	C20—C21—C22	119.0 (3)
C8—C7—H7	119.2	C20—C21—H16	120.5
C6—C7—H7	119.2	C22—C21—H16	120.5
C7—C8—C9	117.5 (3)	C23—C22—C21	120.3 (3)
C7—C8—H8	121.3	C23—C22—H17	119.9
C9—C8—H8	121.3	C21—C22—H17	119.9
C8—C9—N1	130.7 (2)	C22—C23—C18	121.2 (3)
C8—C9—C4	121.8 (2)	C22—C23—H18	119.4
N1—C9—C4	107.5 (2)	C18—C23—H18	119.4

Fig. 1

